Element	With 1 oxygen	With 2 oxygens	With 3	With 4	With 1 hydrogen (and	With 2 hydrogens (and
			oxygens	oxygen	Oxygen)	oxygen
S		$\mathrm{SO_2}^{2-}$	SO_3^{2-}	SO_4^{2-}	HSO ₄ ¹⁻	
		hyposulfite	sulfite	sulfate	hydrogen sulfate	
					(bisulfate)	
P		PO_2^{3-}	PO ₃ ³⁻	PO_4^{-3}	HPO ₄ ²⁻	$H_2PO_4^{1-}$
		hypo phosph ite	phosphite	phosphate	hydrogen phosphate	dihydrogen phosphate
As			AsO_3^{3-}	AsO_4^{-3}		
			arsenite	arsenate		
N		NO_2^{1-}	NO_3^{1-}			
		nitr ite	nitrate			
С			CO_3^{2-}		HCO ₃ ¹⁻	
			carbonate		hydrogen carbonate	
					(bicarbonate)	

For polyatomics for Group 7A elements with oxygen (F, Cl, Br, I)

Element	With 1 oxygen	With 2 oxygen	With 3 oxygens	With 4 oxygens	
In general	Hypoite	Hypoiteiteate		per ate	
F	FO ¹⁻ hypo fluorite	FO ₂ ¹⁻ fluor ite	FO ₃ ¹⁻ fluorate	FO ₄ ¹⁻ per fluorate	
Cl	ClO ¹⁻ hypo chlorite	ClO ₂ ¹⁻ chlor ite	ClO ₃ ¹⁻ chlorate	ClO ₄ ¹⁻ perchlorate	
Br	BrO ¹⁻ hypo brom ite	BrO ₂ ¹⁻ brom ite	BrO ₃ ¹⁻ bromate	BrO ₄ ¹⁻ perbromate	
Ι	IO ¹⁻ hypo iod ite	IO ₂ ¹⁻ iod ite	IO ₃ ¹ iodate	IO ₄ ¹ periodate	

Other ions without a pattern:

Other rons	williout a pattern.					
$\mathrm{NH_4}^{1+}$	ammonium	MnO_4^{1-}	permanganate	$C_2H_3O_2^{1-}$	acetate	$O_2^{2^2}$ peroxide
$H_{3}O^{1+}$	hydromium	MnO_4 ²⁻	manganate	$C_2O_4^{2-}$	oxalate	
OH^{1-}	hydroxide	CrO ₄ ²⁻	chromate	$S_2O_3^{2-}$	thiosulfate	
CN^{1-}	cyanide	$\operatorname{Cr_2O_7}^{2-}$	dichromate	SCN ¹⁻	thiocyanate	